

Raman spectroscopy of newberyite, hannayite and struvite.

Ray L. Frost^{*} ^a, Matt L. Weier ^a, Wayne N. Martens, ^a Dermot A. Henry ^b,
and Stuart J. Mills ^{b,c}

^a Inorganic Materials Research Program, School of Physical and Chemical Sciences, Queensland University of Technology, GPO Box 2434, Brisbane Queensland 4001, Australia.

^b Geosciences, Museum Victoria, PO Box 666E, Melbourne, Victoria 3001, Australia.

^c CSIRO Minerals, Box 312, Clayton South, Victoria 3169, Australia.

This is the authors' version of a paper that was later published as:

Frost, Ray, Weier, Matt, Martens, Wayne, Henry, Dermot & Mills, Stuart (2005) Raman spectroscopy of newberyite, hannayite and struvite. *Spectrochimica Acta* 62(1):pp. 181-188.

Copyright 2005 Elsevier

Abstract

The phosphate minerals hannayite, newberyite and struvite have been studied by Raman spectroscopy using a thermal stage. Hannayite and newberyite are characterised by an intense band at around 980 cm⁻¹ assigned to the ν_1 symmetric stretching vibration of the HPO₄ units. In contrast the symmetric stretching mode is observed at 942 cm⁻¹ for struvite. The Raman spectra are characterised by multiple ν_3 antisymmetric stretching bands and ν_2 and ν_4 bending modes indicating strong distortion of the HPO₄ and PO₄ units. Hannayite and newberyite are defined by bands at 3382 and 3350 cm⁻¹ attributed to HOPO₃ vibrations and hannayite and struvite by bands at 2990, 2973 and 2874 assigned to NH₄⁺ bands. Raman spectroscopy has proven most useful for the analysis of these 'cave' minerals where complex paragenetic relationships exist between the minerals.

Keywords: hannayite, newberryite, struvite, phosphate, Raman spectroscopy

Introduction

Interest in struvite formation also comes from the formation in urinary tracts and kidneys [1-6]. Indeed the discovery of newberyite have been found in very old and large calculi [7]. Newberyite is found as a decomposition product of ivory. More often than not the presence of struvite has been determined by infrared spectroscopy [8-16]. On occasions Raman spectroscopy has also been used to study the presence of struvite in urine [9, 17, 18]. The mineral struvite has the formula (NH₄MgPO₄·6H₂O) and is orthorhombic [19]. The mineral is related to dittmarite (NH₄MgPO₄·H₂O),

^{*} Author to whom correspondence should be addressed (r.frost@qut.edu.au)

niahite $(\text{NH}_4(\text{Mn,Mg})\text{PO}_4 \cdot 6\text{H}_2\text{O})$, hannayite $(\text{NH}_4)_2\text{Mg}(\text{HPO}_4)_4 \cdot 6\text{H}_2\text{O}$, schertelite $(\text{NH}_4)_2\text{MgH}_2(\text{PO}_4)_4 \cdot 4\text{H}_2\text{O}$, stercorite $(\text{Na}(\text{NH}_4)_2\text{HPO}_4 \cdot 4\text{H}_2\text{O})$, swaknoite $(\text{Ca}(\text{NH}_4)_2(\text{HPO}_4)_2 \cdot \text{H}_2\text{O})$, and mundrabillaite $((\text{NH}_4)_2\text{CaHPO}_4 \cdot \text{H}_2\text{O})$. Many of these minerals are found in caves and are the result of formation from guano [20-24].

Often infrared spectroscopy is used to study these minerals in the urinary tracts and calculi [9, 25-28]. The amount of published data on the Raman spectra of mineral phosphates is limited [29-33]. The Raman spectra of the hydrated hydroxy phosphate minerals have not been reported. In aqueous systems, Raman spectra of phosphate oxyanions show a symmetric stretching mode (ν_1) at 938 cm^{-1} , the antisymmetric stretching mode (ν_3) at 1017 cm^{-1} , the symmetric bending mode (ν_2) at 420 cm^{-1} and the ν_4 mode at 567 cm^{-1} [31, 32, 34]. S.D. Ross in Farmer (1974) (page 404) listed some well-known minerals containing phosphate which were either hydrated or hydroxylated or both [35]. The value for the ν_1 symmetric stretching vibration of PO_4 units as determined by infrared spectroscopy was given as 930 cm^{-1} (augelite), 940 cm^{-1} (wavellite), 970 cm^{-1} (rockbridgeite), 995 cm^{-1} (dufrenite) and 965 cm^{-1} (beraunite). The position of the symmetric stretching vibration is mineral dependent and a function of the cation and crystal structure. The fact that the symmetric stretching mode is observed in the infrared spectrum affirms a reduction in symmetry of the PO_4 units. Some studies have been undertaken on $\text{MgHPO}_4 \cdot 3\text{H}_2\text{O}$ [36, 37]. It is not known whether a mineral was studied or whether a synthetic chemical was used. Infrared bands were found at 1162 , 1057 and 1017 cm^{-1} and were assigned to the ν_3 antisymmetric stretching modes. Some more recent studies of newberyite have been undertaken [26, 38].

The value for the ν_2 symmetric bending vibration of PO_4 units as determined by infrared spectroscopy was given as 438 cm^{-1} (augelite), 452 cm^{-1} (wavellite), 440 and 415 cm^{-1} (rockbridgeite), 455 , 435 and 415 cm^{-1} (dufrenite) and 470 and 450 cm^{-1} (beraunite). The observation of multiple bending modes provides an indication of symmetry reduction of the PO_4 units. This symmetry reduction is also observed through the ν_3 antisymmetric stretching vibrations. Augelite shows infrared bands at 1205 , 1155 , 1079 and 1015 cm^{-1} ; wavellite at 1145 , 1102 , 1062 and 1025 cm^{-1} ; rockbridgeite at 1145 , 1060 and 1030 cm^{-1} ; dufrenite at 1135 , 1070 and 1032 cm^{-1} ; beraunite at 1150 , 1100 , 1076 and 1035 cm^{-1} . Some Raman spectra have been published [26, 27]. The studies of the synthetic newberyite showed bands at 398 cm^{-1} assigned to the ν_2 bending mode and at 554 and 509 cm^{-1} corresponding to the ν_4 bending modes [37, 39].

The detailed comparative Raman spectra of the minerals newberyite, hannayite and struvite have not been published. Often the hydroxyl and NH_4^+ stretching vibrations are not given and only the PO_4 or HPO_4 vibrational modes are given. This paper reports the vibrational spectra of these minerals and related the spectra to the known crystal structures.

Experimental

Minerals

Two newberyite samples were obtained from Museum Victoria, samples M25186 and M 14824. These minerals originated from Lava Cave, 5 km S of Skipton (37 44 30 S, 143 21 30 E), Skipton, Victoria, Australia. The mineral hannayite (sample M 13417) was obtained from the same place. The minerals were analysed for phase purity by X-ray diffraction and the compositions were checked by EDX analyses.

Raman microprobe spectroscopy

Samples of the minerals were placed and orientated on the stage of an Olympus BHSM microscope, equipped with 10x and 50x objective lenses, as part of a Renishaw 1000 Raman microscope system. This system also includes a monochromator, filter system and a Charge Coupled Device (CCD). Raman spectra were excited by a HeNe laser (633 nm) at a resolution of 2 cm^{-1} in the range between 100 and 4000 cm^{-1} . Repeated acquisition using the highest magnification was undertaken to improve the signal-to-noise ratio. Spectra were calibrated using the 520.5 cm^{-1} line of a silicon wafer. In order to ensure that the correct spectra were obtained, the incident excitation radiation was scrambled. Previous studies provide an in depth account of the experimental technique [31, 40-44]. Spectral manipulation such as baseline adjustment, smoothing and normalisation was performed using the GRAMS® software package (Galactic Industries Corporation, Salem, NH, USA).

Infrared absorption spectroscopy

Infrared spectra were obtained using a Nicolet Nexus 870 FTIR spectrometer with a smart endurance single bounce diamond ATR cell. Spectra over the $4000\text{--}525\text{ cm}^{-1}$ range were obtained by the co-addition of 64 scans with a resolution of 4 cm^{-1} and a mirror velocity of 0.6329 cm/s .

Results and discussion

The Raman spectra of hannayite and newberyite are shown in a set of figures according to the wavenumber region. Figure 1 displays the hydroxy and NH_4 stretching region of hannayite; Figure 2 the OH and NH_4 stretching region of newberyite and struvite; Figure 3 the $800\text{ to }2000\text{ cm}^{-1}$ region of hannayite; Figure 4 the $1300\text{ to }2000\text{ cm}^{-1}$ region of newberyite and struvite; Figure 5 shows the low wavenumber region of hannayite and Figure 6 the low wavenumber region of struvite and hannayite. The analyses of the Raman spectral data are collected in Table 1 (hannayite) and in Table 2 (struvite and newberyite).

The Raman spectrum of hannayite at 298 K shows a series of bands at 3496, 3460, 3384, 3314, 3219, 3185, 3125, 3090, 2983 and 2872 cm^{-1} . These bands are attributed to a mixture of the hydroxyl and NH_4 stretching vibrations. Further hannayite is a mineral which contains HPO_4 units and hence there will be HOPO_3 vibrations. The Raman spectrum of hannayite is more well defined at 77 K. Raman bands are observed at 3382, 3350, 3217, 3179, 3132, 3120, 2990, 2973 and 2874 cm^{-1} .

In comparison the infrared spectrum shows a spectral profile made up of a series of overlapping bands. According to Catti and Franchini-Angela the structure of hannayite consists of layers of P and Mg coordination polyhedra sharing vertices [45]. Layers are linked by H bonds of H₂O and NH₄⁺ ions, and are formed by 2 crossing systems of [001] double chains and [111] single chains of polyhedra. A distorted coordination is shown by the W(1) water molecule [45]. The NH₄⁺ ion has a coordination no. of six, with 4 ordered H bonds tetrahedrally arranged and 2 electrostatic bonds. The 77 K Raman spectrum may be divided into three sets of bands. These are (a) the bands at 3382 and 3350 cm⁻¹ attributed to HOPO₃ vibrations, (b) bands in the 3120 to 3217 cm⁻¹ region ascribed to water stretching bands, and (c) the bands at 2990, 2973 and 2874 assigned to NH₄⁺ bands. The mineral newberyite does not have any NH₄⁺ units. No bands are observed in the Raman spectra of newberyite between 2800 and 3150 cm⁻¹. This observation confirms the bands for hannayite in this region to NH₄⁺ vibrations.

The Raman spectrum of newberyite shows a series of bands at 3514, 3479, 3456, 3381, 3265 and 3180 cm⁻¹ at 298 K and 3501, 3468, 3364, 3244 and 3277 cm⁻¹ at 77 K. The crystal structure of newberyite has been solved [46]. The structure is orthorhombic with 8 formula units per unit cell. The structure is strongly distorted. In the 77 K spectrum the two bands at 3501 and 3468 cm⁻¹ are assigned to the –POH vibrations in line with the assignment of the bands for hannayite. The bands at 3364, 3244 and 3227 cm⁻¹ are assigned to the water stretching vibrations. There are no bands attributable to NH₄ units in newberyite.

The Raman and IR spectra of hannayite are shown in Figure 3. The Raman spectra are dominated by a single intense band at 974 cm⁻¹ in the 298 K spectrum and at 272 in the 77 K spectrum. The band is sharp with bandwidths of 11.2 and 6.2 cm⁻¹ respectively. This band is attributed to the HPO₄ symmetric stretching vibration. In the infrared spectrum two bands are observed at 1009 and 984 cm⁻¹ and are attributed to the infrared activated PO₄ symmetric stretching vibration. Normally such a vibration should not be observed but because of the distortion of the PO₄ units in the hannayite structure the bands become infrared-active. For newberyite, at 298 K an intense band attributed to the PO₄ symmetric stretching vibration is observed at 984 cm⁻¹ with band width of 12.1 cm⁻¹. The band is found at 984 cm⁻¹ with a bandwidth of 7.1 cm⁻¹ at 77 K. In the infrared spectrum two bands are observed at 994 and 1016 cm⁻¹. For struvite the PO₄ symmetric stretching band is observed at 942 cm⁻¹. The difference in the band positions for the struvite as compared with the hannayite and newberyite is that struvite contains PO₄ units where as the other two minerals contain HPO₄ units.

In the Raman spectrum of hannayite three bands are observed at 1172, 1070 and 1011 cm⁻¹ at 298 K and at 1179, 1070 and 1012 cm⁻¹. These bands are attributed to the HPO₄ antisymmetric stretching modes. The Raman data is in agreement with published infrared data for synthetic MgHPO₄·3H₂O where IR bands were found at 1162, 1057 and 1017 cm⁻¹. Our ATR spectrum of hannayite is a complex set of overlapping bands. Bands are observed at 1179, 1070 and 1009 cm⁻¹ together with other low intensity bands. In the Raman spectrum of newberyite only a low intensity band at 1154 cm⁻¹ is observed (1162 cm⁻¹ at 77 K). In the ATR spectrum three bands are observed at 1161, 1071, and 1055 cm⁻¹. These bands are attributed to the HPO₄

antisymmetric stretching modes. For struvite only low intensity Raman bands at 1076 and 1013 cm^{-1} are observed.

The Raman spectra of hannayite in the low wavenumber region are shown in Figure 5. A band is observed in the 298 K spectrum at 556 cm^{-1} of half width of 36.1 cm^{-1} which splits into two bands at 574 and 555 cm^{-1} at 77 K of half width 11.1 and 8.8 cm^{-1} . These bands are attributed to the PO_4 ν_4 bending mode. Published spectra of newberyite gave the two bands at 554 and 509 cm^{-1} as the ν_4 bending modes [37, 39]. A Raman band was observed at 522 cm^{-1} which shifted to 518 cm^{-1} at 77 K which may also be ascribed to a the ν_4 bending mode. In the infrared spectrum a broad band is observed at 531 cm^{-1} which may be the corresponding infrared band. In the Raman spectrum of newberyite, a broad band is observed at 555 cm^{-1} which appears to split into four bands at 557, 537, 522 and 504 cm^{-1} at 77 K. The observation of multiple ν_4 bending modes is a strong indication of the reduction of symmetry of the HPO_4 ion. For struvite the ν_4 bending mode is observed at 564 cm^{-1} . Two bands are observed in the ATR spectrum at 567 and 552 cm^{-1} which may be assigned to this vibration.

In the Raman spectrum of hannayite Raman bands are observed at 436, 415, and 375 cm^{-1} which may be assigned to the PO_4 ν_2 bending modes. These bands become more clearly resolved in the 77 K spectrum where bands are observed at 424, 403 and 385 cm^{-1} . bands in these positions are not observed in the ATR spectrum as the bands are below the limit of the ATR method. The Raman spectrum of newberyite shows an intense band at 399.8 cm^{-1} . At 77 K the band splits into three bands at 409, 400 and 384 cm^{-1} . The mineral struvite's Raman spectrum displays two bands at 463 and 428 cm^{-1} which can be assigned to this vibrational mode.

In the Raman spectrum of hannayite at 77 K a series of bands are found at 277, 254, 237, 221 and 197 cm^{-1} . The first band is of significant intensity and is attributed to the MgO stretching vibration. The band at 197 cm^{-1} may be due to the OMgO bending mode. For newberyite intense bands in the 298 K spectrum are observed at 283 and 219 cm^{-1} . The Raman spectrum of struvite also shows an intense band at 300 cm^{-1} which is attributed to the MgO stretching vibration.

Conclusions

The minerals hannayite, newberyite and struvite may be termed 'cave' minerals. These mineral formed through deposits on walls of bat excrement have been analysed by Raman spectroscopy in combination with a thermal stage. The minerals hannayite and newberyite contain hydrogen phosphate units (HPO_4) where as struvite is based upon phosphate units (PO_4). The HPO_4 and PO_4 units are characterised by intense sharp bands at 980 cm^{-1} and 942 cm^{-1} . Characteristic POH stretching bands are observed at 3382 and 3350 cm^{-1} . Such bands are absent in the Raman spectrum of struvite. Hannayite and struvite contain NH_4^+ units which are observed by NH stretching vibrations at 2990, 2973 and 2874. The use of a portable Raman spectrometer would enable the identification of such 'cave' minerals. In particular where close paragenetic relationships exist between these minerals Raman spectroscopy enables the identification and study of such minerals.

Acknowledgements

The financial and infrastructure support of the Queensland University of Technology Inorganic Materials Research Program of the School of Physical and Chemical Sciences is gratefully acknowledged. The Australian Research Council (ARC) is thanked for funding. Museum Victoria is thanked for the loan of the minerals. SJM wishes to thank the support of CSIRO Minerals.

References

- [1]. H. J. Schneider and M. Anke, *Urologia* 24 (1969) 300.
- [2]. H. J. Schneider, L. Klotz and G. Horn, *Urologia* 62 (1969) 351.
- [3]. H. J. Schneider, *Urologia* 62 (1969) 123.
- [4]. H. J. Schneider and G. Horn, *Urologia* 61 (1968) 753.
- [5]. H. J. Schneider and M. Anke, *Urologia* 61 (1968) 361.
- [6]. H. J. Schneider and M. Anke, *Urologia* 61 (1968) 361.
- [7]. K. Lonsdale and D. J. Sutor, *Science* 154 (1966) 1353.
- [8]. M. Daudon, C. Marfisi, B. Lacour and C. Bader, *Clin. Chem.* 37 (1991) 83.
- [9]. M. Daudon, M. F. Protat, R. J. Reveillaud and H. Jaeschke-Boyer, *Kidney International* 23 (1983) 842.
- [10]. E. Escolar and J. Bellanato, *Biospec.* 5 (1999) 237.
- [11]. M. H. Gault, M. Ahmed, J. Kalra, I. Senciall, J. Morgan, W. Cohen and D. Churchill, *Urolithiasis: Clinical Basic Res., [Proc. Int. Symp.]*, 4th (1981) 993.
- [12]. A. Hesse, M. Gergeleit, P. Schueller and K. Moeller, *J. Clin. Chem. Clin. Biochem.* 27 (1989) 639.
- [13]. A. Khaliq, J. Ahmed and N. Khalid, *British J. Urol.* 56 (1984) 135.
- [14]. W. E. Klee, *Fortschritte der Urologie und Nephrologie* 9 (1977) 234.
- [15]. Y. G. Kozlovsky, M. M. Shokarev and F. I. Vershinina, *Laboratornoe Delo* (1977) 659.
- [16]. C. Paluszkievicz, M. Galka, W. Kwiatek, A. Parczewski and S. Walas, *Biospec.* 3 (1997) 403.
- [17]. K. Sudlow and A. Woolf, *Clinica Chimica Acta* 203 (1991) 387.
- [18]. K. Angoni, J. Popp and W. Kiefer, *Spec. Letts.* 31 (1998) 1771.
- [19]. J. A. Bland and S. J. Basinski, *Nature* 183 (1959) 1385.
- [20]. P. J. Bridge, *Min. Mag.* 39 (1973) 467.
- [21]. P. J. Bridge, *Min. Mag.* 41 (1977) 33.
- [22]. P. J. Bridge and R. M. Clark, *Min. Mag.* 47 (1983) 80.
- [23]. P. J. Bridge and B. W. Robinson, *Min. Mag.* 47 (1983) 79.
- [24]. R. J. Bowell, A. Warren and I. Redmond, *Geol. Soc. Special Publication* 113 (1996) 63.
- [25]. M. Modlin and P. J. Davies, *South African Medical Journal* 59 (1981) 337.
- [26]. B. Soptrajanov, G. Jovanovski, I. Kuzmanovski and V. Stefov, *Spec. Letts* 31 (1998) 1191.
- [27]. B. Soptrajanov, I. Kuzmanovski, V. Stefov and G. Jovanovski, *Spec. Letts* 32 (1999) 703.
- [28]. M. Volmer, J. C. M. De Vries and H. M. J. Goldschmidt, *Clin. Chem.* 47 (2001) 1287.
- [29]. R. L. Frost, L. Duong and W. Martens, *Neues Jahrbuch fuer Mineralogie, Monatshefte* (2003) 223.
- [30]. R. L. Frost, W. Martens, P. A. Williams and J. T. Klopogge, *J. Ram. Spec.* 34 (2003) 751.
- [31]. R. L. Frost, W. Martens, P. A. Williams and J. T. Klopogge, *Min. Mag.* 66 (2002) 1063.
- [32]. R. L. Frost, P. A. Williams, W. Martens, J. T. Klopogge and P. Leverett, *J. Ram. Spec.* 33 (2002) 260.

- [33]. R. L. Frost, P. A. Williams, W. Martens and J. T. Klopogge, *J. Ram. Spec.* 33 (2002) 752.
- [34]. R. L. Frost, W. N. Martens, T. Klopogge and P. A. Williams, *Neues Jahrbuch fuer Mineralogie, Monatshefte* (2002) 481.
- [35]. V. C. Farmer, *Mineralogical Society Monograph 4: The Infrared Spectra of Minerals*, 1974.
- [36]. F. A. Miller, G. L. Carlson, F. F. Bentley and W. H. Jones, *Spectrochim. Acta* 16 (1960) 135.
- [37]. F. A. Miller and C. H. Wilkins, *Anal. Chem.* 24 (1952) 1253.
- [38]. I. L. Shashkova, N. V. Kitikova, A. I. Rat'ko and A. G. D'yachenko, *Inorg. Mats. (Translation of Neorganicheskie Materialy)* 36 (2000) 826.
- [39]. J. R. Lehr, E. H. Brown, A. W. Frazier, J. P. Smith and R. D. Thrasher, *Tennessee Valley Authority, Chemical Engineering Bulletin No. 6* (1967) 166 pp.
- [40]. R. L. Frost, M. Crane, P. A. Williams and J. T. Klopogge, *J. Ram. Spec.* 34 (2003) 214.
- [41]. R. L. Frost, P. A. Williams and W. Martens, *Min. Mag.* 67 (2003) 103.
- [42]. W. Martens, R. L. Frost and J. T. Klopogge, *J. Ram. Spec.* 34 (2003) 90.
- [43]. W. Martens, R. L. Frost, J. T. Klopogge and P. A. Williams, *J. Ram. Spec.* 34 (2003) 145.
- [44]. R. L. Frost, W. Martens, J. T. Klopogge and P. A. Williams, *J. Ram. Spec.* 33 (2002) 801.
- [45]. M. Catti and M. Franchini-Angela, *Acta Cryst. B* 32 (1976) 2842.
- [46]. D. J. Sutor, *Acta Cryst.* 23 (1967) 418.

Table 1 Results of the vibrational spectroscopy of hannayite

ATR			298K Raman			77K Raman		
Center	FWHM	Area	Center	FWHM	Area	Center	FWHM	Area
3490	29.7	0.005	3496	26.6	0.006			
3483	103.2	0.046						
			3460	118.7	0.030			
3424	36.8	0.002				3382	39.8	0.020
3373	89.1	0.081	3384	70.7	0.024	3350	19.4	0.058
			3314	79.6	0.015			
3288	95.7	0.032	3219	109.2	0.049	3217	52.9	0.066
3199	137.9	0.135	3185	44.5	0.017	3179	37.1	0.103
			3125	40.1	0.021	3132	14.4	0.018
3111	39.8	0.006				3120	35.5	0.093
3066	171.0	0.143	3090	105.0	0.029			
			2983	119.0	0.037	2990	43.1	0.059
2966	100.4	0.044	2872	61.8	0.015	2973	14.1	0.035
2875	19.5	0.001				2874	18.7	0.065
2865	59.7	0.011						
2827	193.4	0.102						
2680	108.6	0.011						
			2649	122.0	0.012			
2456	70.6	0.007	2466	209.5	0.045			
2392	49.3	0.005						
2306	222.6	0.029						
			2228	100.0	0.005			
			1947	170.0	0.044			
1890	36.1	0.001						
1812	95.7	0.005	1756	302.4	0.149			
1698	41.3	0.004	1708	21.7	0.003	1713	9.0	0.014
1652	42.2	0.007	1661	85.5	0.009			
1635	181.0	0.026						
1486	81.1	0.002	1457	21.1	0.001	1459	25.1	0.021
1459	11.7	0.002						
1448	31.2	0.016	1429	136.2	0.073			
1444	11.3	0.002						
1415	40.9	0.004						
1294	31.3	0.007						
1250	32.2	0.005	1227	67.7	0.012			
			1172	73.3	0.022	1179	4.3	0.007
1179	31.9	0.017						
1158	16.6	0.002	1119	23.1	0.001			
1142	18.6	0.010	1070	12.6	0.002	1070	6.6	0.007
						1067	31.0	0.021
1070	55.4	0.064	1011	11.4	0.002	1012	6.2	0.021
1056	21.0	0.003				999	8.5	0.005
1009	23.3	0.012				972	6.2	0.161
984	28.8	0.037	974	11.2	0.026			
			971	92.8	0.029			
965	49.8	0.020						

924	33.1	0.004						
890	11.6	0.002	882	14.4	0.001	887	14.7	0.024
880	27.4	0.013						
848	49.7	0.007				819	21.5	0.016
			802	52.6	0.008			
765	90.1	0.018	757	63.6	0.014	611	12.5	0.004
636	55.3	0.007	596	73.8	0.020	574	11.1	0.012
581	18.0	0.001	556	36.1	0.008	555	8.8	0.018
531	77.9	0.002	522	8.9	0.003	518	16.6	0.007
			513	112.7	0.018			
			436	14.4	0.001	424	24.5	0.021
			415	26.4	0.002	403	7.3	0.010
			379	155.9	0.052	385	19.8	0.040
			375	15.2	0.001			
			356	27.0	0.001	277	9.2	0.021
			269	9.6	0.000	254	9.8	0.002
			250	7.4	0.001	237	3.9	0.006
			247	57.1	0.018	221	4.8	0.008
			205	22.5	0.007	197	11.1	0.015
			193	16.2	0.003	178	7.5	0.006
						157	7.3	0.002

Table 2

Struvite ATR			Struvite 298K Raman			Newberyite ATR			Newberyite 298K Raman			Newberyite 77K Raman		
Center	FWHM	Area	Center	FWHM	Area	Center	FWHM	Area	Center	FWHM	Area	Center	FWHM	Area
3583.5	61.1	0.002				3535.9	39.8	0.002				3501.2	13.1	0.031
						3515.1	23.3	0.004						
						3488.7	128.1	0.080				3514.0	48.0	0.029
3471.8	142.2	0.010				3474.3	22.5	0.003				3468.5	12.3	0.082
												3479.3	26.6	0.080
												3456.0	62.1	0.073
						3376.7	78.8	0.041				3381.1	92.3	0.153
						3263.8	92.5	0.089				3265.4	85.1	0.135
												3364.0	27.2	0.018
						3216.9	265.9	0.214				3244.5	42.6	0.193
3169.7	294.2	0.069	3239.3	95.8	0.023				3180.8	80.3	0.019	3227.4	20.6	0.054
			3115.2	204.5	0.181									
			2921.4	143.3	0.083									
2867.7	400.6	0.530	2902.8	6.8	0.001	2866.2	329.9	0.119	2879.6	160.2	0.050	2384.4	54.0	0.029
2507.6	313.5	0.073												
						2474.4	43.2	0.004						
						2401.8	78.8	0.010						
			2367.6	135.1	0.041									
2331.8	163.4	0.059				2322.3	113.8	0.012						
1803.0	179.8	0.010												
						1697.5	44.5	0.014						
1674.8	60.0	0.005				1648.3	31.5	0.024	1648.3	23.3	0.015	1669.2	52.4	0.039
						1625.4	113.6	0.016	1619.8	93.8	0.030	1649.7	10.9	0.009
1591.1	145.8	0.043												
1440.4	58.9	0.006				1495.9	150.6	0.010						

1065.4	40.9	0.003	1076.5	15.7	0.017	1234.5	27.7	0.014	1271.8	29.6	0.010	1279.4	17.9	0.006			
						1161.3	31.9	0.045	1194.9	20.2	0.004						
									1154.0	14.3	0.018	1161.7	7.8	0.024			
												1100.8	71.7	0.023			
									1071.4	53.2	0.031						
									1054.9	24.4	0.020						
									1016.0	32.0	0.049						
									993.7	65.9	0.033	983.8	12.1	0.089			
												967.1	53.8	0.026			
															983.9	7.1	0.197
980.4	72.5	0.074				949.5	18.2	0.042									
						941.8	25.4	0.352						967.1	53.8	0.026	
893.7	123.1	0.055	889.6	50.5	0.015	900.9	14.9	0.002				908.1	19.2	0.055			
						881.5	39.0	0.032							892.6	22.9	0.073
747.7	121.8	0.019				747.6	59.6	0.007									
678.5	69.0	0.005				687.5	27.0	0.001				714.2	20.2	0.005			
						655.9	151.9	0.101							679.8	116.9	0.041
						596.9	30.8	0.002									
567.1	23.0	0.003	564.1	15.5	0.094												
551.6	29.8	0.004				548.2	25.0	0.005	554.8	14.6	0.005	557.2	9.4	0.005			
												537.0	14.8	0.015			
						524.6	49.9	0.000				522.5	11.8	0.006			
									498.4	20.0	0.032	503.9	5.1	0.009			
			462.8	22.7	0.011												
			427.5	22.9	0.016												
												409.0	11.0	0.037			
									399.8	12.7	0.076	400.1	7.4	0.023			
									368.7	17.0	0.009	384.4	17.3	0.048			
									327.0	8.6	0.001	335.2	8.2	0.009			
												312.5	3.1	0.001			
			299.6	22.2	0.060							310.3	5.9	0.004			
									282.9	14.4	0.006	287.9	5.8	0.004			
									265.7	19.3	0.003	257.4	8.9	0.007			

	242.1	13.6	0.016		244.4	8.5	0.002	242.0	12.5	0.005
	228.5	11.5	0.015		219.3	10.9	0.011	219.5	9.9	0.007
	205.6	12.5	0.024		199.4	8.3	0.004	198.9	10.0	0.010
					180.0	9.7	0.006	177.9	10.4	0.011
					158.4	4.9	0.000	157.9	3.9	0.003
					138.5	5.2	0.001			

List of Figures

Figure 1 Raman spectra at 298 and 77 K and the ATR spectrum of the OH and NH stretching region of hannayite.

Figure 2 Raman spectra at 298 and 77 K and the ATR spectrum of the OH and NH stretching region of newberyite and struvite.

Figure 3 Raman spectra at 298 and 77 K and the ATR spectrum of the 800 to 2000 cm^{-1} region of hannayite.

Figure 4 Raman spectra at 298 and 77 K and the ATR spectrum of the 100 to 1300 cm^{-1} region of newberyite and struvite.

Figure 5 Raman spectra at 298 and 77 K and the ATR spectrum of the 100 to 800 cm^{-1} region of hannayite.

Figure 6 Raman spectra at 298 and 77 K and the ATR spectrum of the 100 to 800 cm^{-1} region of newberyite and struvite.

List of Tables

Table 1 Results of the spectral analysis of the Raman and infrared spectrum of hannayite

Table 2 Results of the spectral analysis of the Raman and infrared spectrum of newberyite and struvite

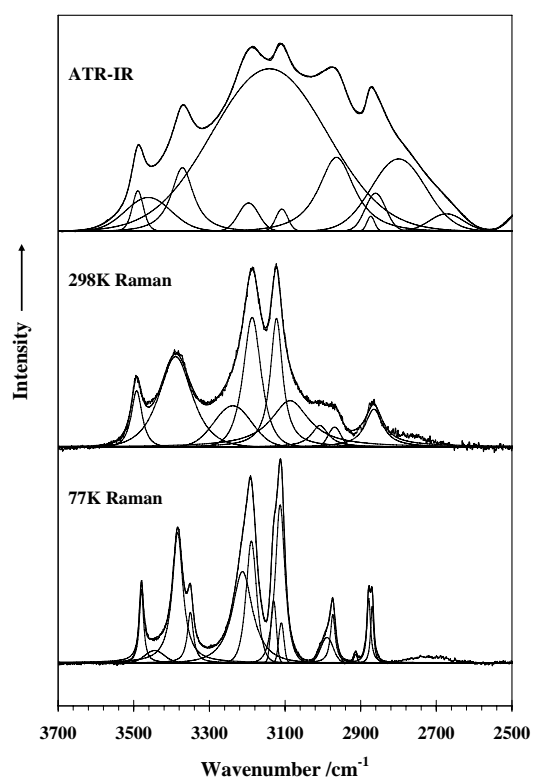


Figure 1 Hannayite

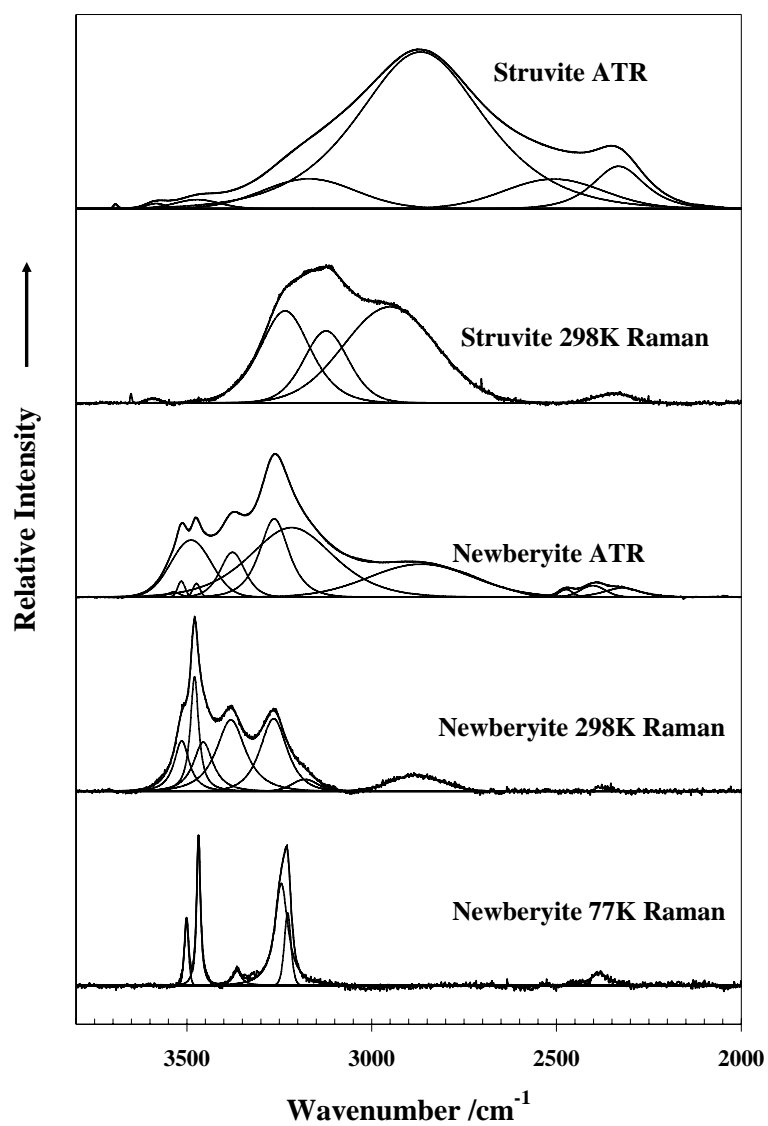
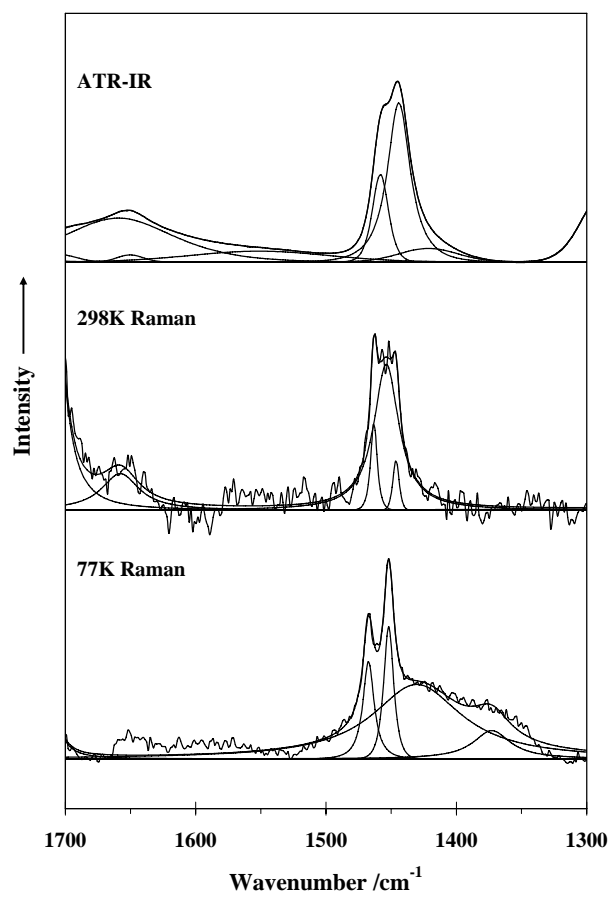


Figure 2



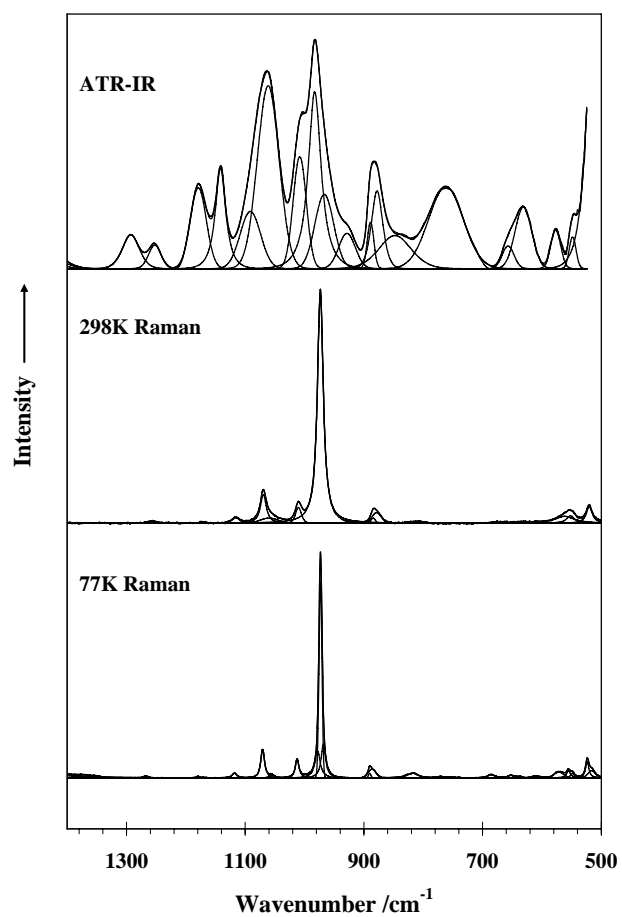


Figure 3 Hannayite

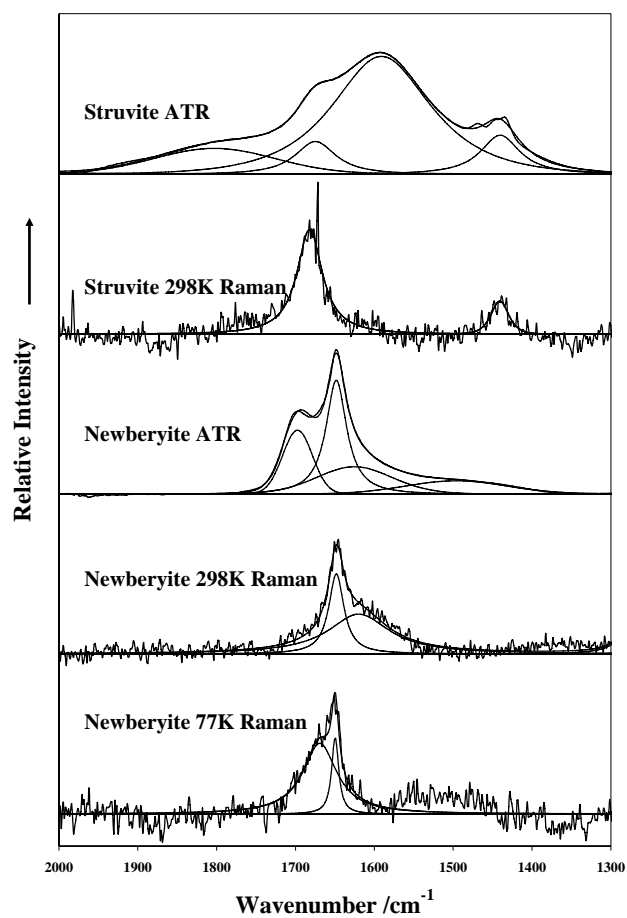


Figure 4 Newberyite

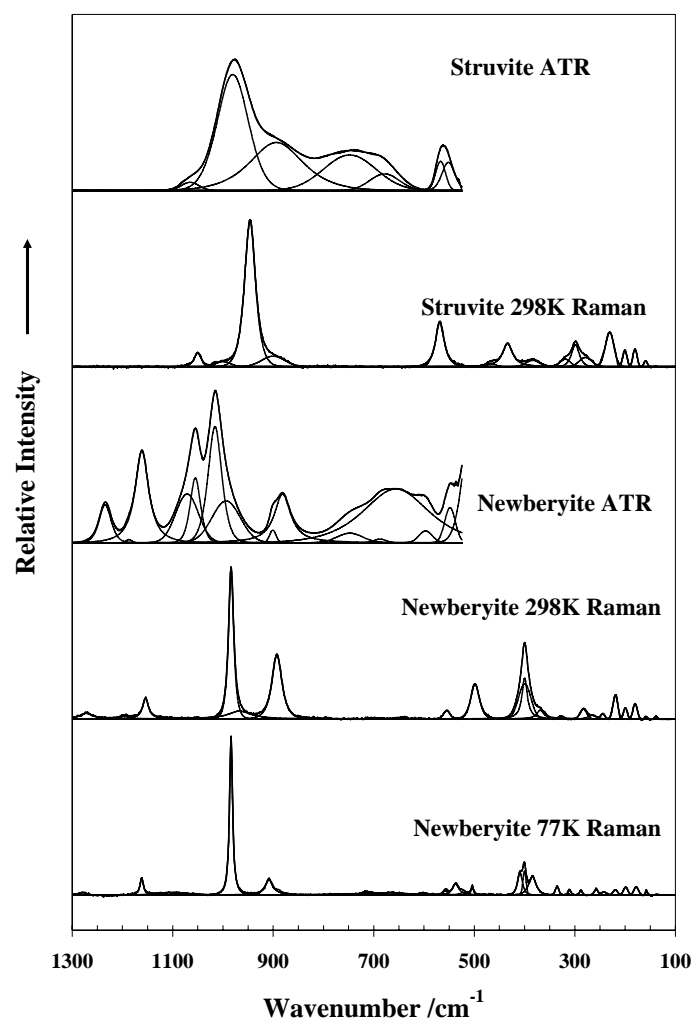


Figure 6

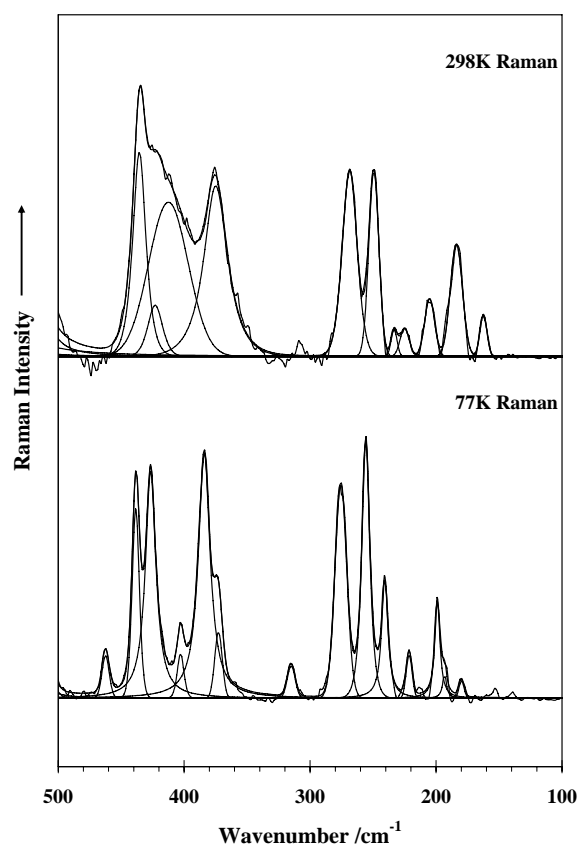


Figure 5 Hannayite